## ASYMMETRIC SYNTHESIS XX<sup>1</sup>. PREPARATION OF A NOVEL SPIROPIPERIDINE SYSTEM BY THE CN(R,S) METHOD.

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<u>Abstract</u>: A novel spiropiperidine system i.e an aza structural analog of histrionicotoxin has been constructed from 2-cyano-6-phenyloxazolopiperidine synthon  $\underline{3}$ . The suitable quaternary carbon was created by alkylation at the position  $\alpha$  to the nitrile followed by transformation of the nitrile group into a primary amine. Final aminoreductive cyclisation gave the spiro system.

Natural histrionicotoxin and its perhydro congeners possess unique pharmacological activity as neurotoxins<sup>2</sup>. Structure activity relationship studies suggested the importance of the two heteroatoms in the mode of action of these series of alkaloids<sup>3</sup>. It was therefore attractive to synthesize a molecule in which two piperidine rings have a spiro-junction i.e. aza structural analogs  $\frac{9}{4}$  of both the hitherto reported spiropiperidine alkaloids histrionicotoxin 1 and nitramine  $\frac{4}{4}$  (figure).

**Figure** 

Our operating strategy was based on our recent finding that 1,2-diamines could be obtained in a stereoselective manner from 2-cyano-6-phenyloxazolopiperidine synthon  $3^5$ . Indeed reaction of a lithio derivative or a cuprate on the nitrile group of 3 afforded an imine which was subsequently reduced to a primary amine. However our strategy involved the reactivity of a tertiary nitrile whose behaviour was unknown; in particular elimination could not be excluded 6.

The synthesis started with the alkylation of synthon  $\frac{3}{2}$  by the ethylene ketal of bromo-3 propanal (Scheme). The absolute configuration at C-2 is known as R on the basis

i) LDA, THF, - 78°C then 2-(2-Bromoethyl)-1,3-dioxolane; ii) BuLi, Et<sub>2</sub>O, - 78°C  $\rightarrow$  0°C; iii) NaBH<sub>4</sub>, MeOH; iv) CH<sub>2</sub>Cl<sub>2</sub>, H<sub>2</sub>O, KCN, HCl, pH. 2-4; v) NaBH<sub>4</sub>, MeOH; vi) H<sub>2</sub>, Pd-C, MeOH; vii) HCHO, CH<sub>3</sub>COOH.

of our previous results  $^7$ . Compound  $^4$  was submitted to reaction with BuLi in ether to give the imine  $^5$  (y = 84%); the formation of elimination or dialkylation products was not observed. The use of cuprate derivative did not improve the yield as observed in our first report  $^5$ . The concomitant reduction of the imine and amino-ether functions were performed with NaBH $_4$  in MeOH; the primary amine  $^6$  was obtained as a mixture of two isomers  $^6$  (95/5) in 94% yield. The absolute configuration of the new chiral center C-9 as R in the major product  $^6$  can be explained by a preferred attack on the Re-face, as discussed earlier  $^5$ , to minimize steric hindrance. The Si-face attack would imply an approach of the nucleophile with three strong steric interactions (H-4, H-6 and H-7) so the absolute configuration of  $^6$  is 2R,9R.

All attempts to obtain the bicyclic product  $\underline{9}$  directly from  $\underline{6a}$  by a deprotective and amino-reductive sequence failed. The only product isolated resulted from the condensation of the aldehyde with the nitrogen of the piperidine ring 10. It was thus

necessary to follow a stepwise scheme. The deprotection of the aldehyde function and cyclization were performed in acidic medium. The intermediate iminium was trapped with KCN to furnish the unstable amino-nitrile  $\frac{7}{2}$  which was not purified but subjected immediately to reduction (NaBH $_4$ , MeOH), leading to the diamino alcohol  $\frac{8}{1}$ . The hydrogenolysis of the chiral appendage furnished the spirodiamine  $\frac{9}{1}$  as a single compound (y = 85% from  $\frac{6}{2}$ ). The absolute configuration of the two chiral centers being settled as R, the examination of Dreiding models showed that  $\frac{9}{2}$  can adopt the conformation indicated on the scheme in which the C-2..N-1 bond is axial and the butyl chain equatorial. In order to confirm this structure we prepared the tricyclic derivative  $\frac{10}{2}$  by condensation of the diamine with formaldehyde in acidic medium. The aminal  $\frac{10}{2}$  was obtained in good yield showing the proximity of the two nitrogen atoms.

In conclusion, the reaction of synthon  $\underline{1}$  with organometallic derivatives is a valuable method for the preparation of 1',2'substituted 1,2-diamines allowing the construction of a new aza-spiro skeleton. The preparation of more functionalized compounds and the study of their pharmacological properties are under investigation and will be reported later.

## References and notes

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- 7 L. Guerrier, J. Royer, D.S. Grierson and H.-P. Husson, J. Am. Chem. Soc., 1983, 105, 7754.
- 8 Diamine <u>6</u>. <u>6a</u> : oil,[ $\alpha$ ]  $\frac{20}{D}$  : 33° (CHCI<sub>3</sub>, c = 3.0). IR : 3300, 2900, 1460cm<sup>-1</sup>. MS, m/e (%) : 390 (M<sup>+</sup>, 2), 359(5), 304(100). <sup>1</sup>H NMR : 0.90 (t, J = 6.0 Hz, Me),

1.2-1.8 (m, 16H), 2.67 (td, J = 9.2 Hz, J = 2.5 Hz, H-6ax), 3.0-3.2 (m, H-6eq, H-9, OH, NH<sub>2</sub>), 3.42 (dd, J = 10.8 Hz, J = 4.9 Hz, H-7 or H-8), 3.70 (m, 4H,  $-OCH_2CH_2O$ -), 3.86 (t, J = 10.8 Hz, H-7 or H-8), 4.50 (br.s, H-12), 4.61 (dd, J = 10.8 Hz, J = 4.9 Hz, H-7 or H-8), 7.2-7.4 (m, 5H). <sup>13</sup>C NMR : 13.9 (Me), 21.1, 22.6, 26.5, 29.1, 29.9, 30.1, 32.0 (8 CH<sub>2</sub>), 41.7 (C-9), 57.8 (C-6), 60.4 (C-7), 61.1 (C-2), 62.7 (C-8), 64.6 (OCH<sub>2</sub>CH<sub>2</sub>O), 104.2 (C-12), 126.8, 127.7, 129.3, 140.2 (ArC).

 $\frac{6b}{(M^{+},2)} : \text{oil, } \mathbb{E}_{Q} \mathbb{I}_{D}^{20} : -44^{\circ} \text{ (CHCl}_{3}, c = 3.0). } \text{ IR} : 3300, 2950cm^{-1}. MS, m/e (%) : 390 (M^{+},2), 359(8), 304(100). } \mathbb{I}_{H} \text{ NMR} : 0.95 (t, J = 6.0 Hz, Me), 1.2-2.0 (m, 16H), 2.57 (td, J = 9.3 Hz, J = 2.5 Hz, H-6ax), 3.23 (ddd, J = 9.3 Hz, J = 2.0 Hz, J = 1.0 Hz, H-6 eq), 3.46 dd, (J = 10.8 Hz, J = 2.3 Hz, H-7 or H-8), 3.70 (m, 4H, OCH_{2}CH_{2}O), 3.72 (m, H-9), 4.01 (t, J = 10.8 Hz, H-7 or H-8), 4.24 (dd, J = 10.8 Hz, J = 2.3 Hz, H-7 or H-8), 7.2-7.4 (m, 5H). <math display="block"> \mathbb{I}_{3}^{1} \text{ C NMR} : 13.8 \text{ (Me)}, 21.1, 22.6, 25.9, 28.4, 28.7, 28.9, 29.3, 30.7 (8 CH_{2}), 42.5 (C-9), 57.7 (C-6), 60.4 (C-7), 61.3 (C-8), 62.2 (C-2), 64.9 (OCH_{2}CH_{2}O), 103.9 (C-12), 127.6, 128.1, 130.0, 137.9 (Ar.C).$ 

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- Detailed results of this reaction will be reported elsewhere.
- Spiro derivative  $\underline{8}$ : IR: 3300,  $2950 \, \mathrm{cm}^{-1}$ . MS, m/e (%): 330 (M<sup>+</sup>, 2), 299 (77), 231(62), 230(100), 217(65), 209(67), 124(69). <sup>1</sup>H NMR: 0.90 (t, J = 6.0 Hz, Me), 1.0–1.8 (m, 16H), 1.90 (m, H–12 ax), 2.67 (m, H–9), 2.75 (m, H–6ax), 2.95 (dt, J = 14.5 Hz, J = 3.9 Hz, H–6 eq), 3.28 (ddd, J = 9.5 Hz, J = 1.5 Hz, J = 1.0 Hz, H–12 eq), 3.43 (dd, J = 10.3 Hz, J = 5.0 Hz, H–7 or H–8), 3.91 (t, J = 10.3 Hz, H–7 or H–8), 4.53 (dd, J = 10.3 Hz, J = 5.0 Hz, H–7 or H–8). <sup>13</sup>C NMR: 14.3 (Me), 19.5, 22.7, 22.9, 23.3, 25.2, 29.4, 31.1, 31.3 (8 CH<sub>2</sub>), 39.2 (C–12 or C–6), 40.2 (C–6 or C–12), 56.1 (C–2), 56.9 (C–9), 59.0 (C–7), 61.7 (C–8).
- 7-butyl 1,8-diazaspiro (5,5)-undecane  $\underline{9}: [\alpha]_{D}^{20}: +40^{\circ} (CHCl_{3}, c=1.0). IR: 3300, 2950, 1420cm^{-1}. MS: m/e (%): 210 (M<sup>+</sup>, 5), 184(100), 153(10), 96(100). <sup>1</sup>H NMR: 0.90 (t, J=6.0 Hz, Me), 1.0-2.0 (m, 15H), 2.28 (m, H-9 ax), 2.38 (dt, J=13.9 Hz, J=2.5 Hz, H-3 eq or H-10 eq), 2.58 (td, J=10.2 Hz, J=4.2 Hz, H-12 ax), 2.77 (m, 2xH-6), 3.00 (dt, J=11.6 Hz, J=3.8 Hz, H-12 eq). <sup>13</sup>C NMR: 14.2 (Me), 19.9, 22.1, 23.0, 26.9, 28.3, 29.3, 30.2, 33.4 (8 CH<sub>2</sub>), 40.2 (C-6), 46.8 (C-12), 52.2 (C-2), 65.2 (C-9).$
- Aminal  $\underline{10}: [\alpha]_D^{20}: +67^{\circ} (CHCl_3, c=1.5).$  IR:  $2950cm^{-1}.$  MS: m/e (%) 222 (M<sup>+</sup>, 30), 181(100), 165(63), 123(53). H NMR: 0.88 (t, J=6.5 Hz, Me), 1.1-2.1 (m, 16H), 2.6-2.9 (m, 5H, 2xH-6, 2xH-10, H-9), 3.58 and 3.87 (2d, 2d,  $N-CH_2-N$ ).  $13^{13}$ C NMR: 14.2 (Me), 19.9, 20.1, 23.1, 25.5, 27.8, 29.8, 30.2, 32.7 (8  $CH_2$ ), 44.7 (C-6 or C-12), 56.3 (C-12 or C-6), 59.0 (C-2), 71.9 (C-9), 74.7 (N- $CH_2$ -N).